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Synthesis of novel FeSiBPCCu alloys with high amorphous forming ability and good soft magnetic properties



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ARTICLEINFO	A B S T R A C T
<i>Keywords:</i> Nanocrystalline alloy Amorphous forming ability Microstructure Saturation magnetic flux density	Novel Fe _{83.3} Si ₄ B ₈ P _{4.x} C _x Cu _{0.7} ($x = 0$ –4) alloys were developed with the aim of increasing amorphous forming ability (AFA) without deteriorating the soft magnetic properties. Microstructure analysis revealed that the substitution of C for P inhibited the precipitation of α -Fe phase during the quenching process hence increased the AFA. The critical thickness increased from 18 µm for the alloys with composition of $x = 0$ to 43 µm for $x = 4$, respectively, owing to the decrease in the melting entropy (ΔS_m), the approach of eutectic point and micro- alloying effect. Saturation magnetic flux density (B_s) increased obviously after proper annealing with increasing C substitution due to the increase in volume fraction of α -Fe nanocrystals, while excessive C content leaded larger coercivity (H_c) but improved the frequency and <i>DC</i> superposition performance. The Fe _{83.3} Si ₄ B ₈ P ₂ C ₂ Cu _{0.7} nanocrystalline alloy was successfully synthesized with high B_s of 1.84 T, low H_c of 4.8 A/m and high effective permeability (μ_c) of 13,540. The combination of high B_s and high resistance to <i>DC</i> bias promises potential

material in DC superposition application.

1. Introduction

Over the past three decades, nanocrystalline soft magnetic alloys of FeSiBNbCu, FeZrB(Cu) and FeCoZrBCu under the trade name of FINEMET, NANOPERM and HITPERM have been successively developed [1–3]. The research and application of nanocrystalline alloys [1–6] have greatly promoted the development of modern power electronic industry. However, with starting the problems of energy shortage and environmental pollution, more requirements are put forward on nanocrystalline alloys, and it becomes a noticeable topic to develop nanocrystalline alloys combined with high saturation magnetic flux density (B_s), low coercivity (H_c), low core loss, as well as low materials cost. The traditional nanocrystalline alloys cannot meet the above requirements anymore.

From 2009, nanocrystalline FeSiBPCu alloy (NANOMET) has attracted great attention due to its high $B_{\rm s}$ of > 1.8 T, low $H_{\rm c}$ of < 10 A/m and low material cost [7,8]. However, the relatively low amorphous forming ability (AFA), by which the ribbon thickness must be suppressed to 20 µm or less, constrains its application. Therefore, how to increase the AFA has become the crucial problem for NANOMET alloy.

It has been reported that flux melting is an effective way to improve the AFA [9,10]. The fluxing purification by B_2O_3 and CaO has been proved feasible to increase the AFA in high B_s FeSiBPC amorphous alloy because it can suppress the surface crystallization and alleviate the effect of detrimental impurity such as oxide and sulfide [11]. But the complex fluxing process increases the manufacturing cost. Recently, it is found that a proper amount of Co substitution for Fe in NANOMETtype alloys is effective to increase the AFA without losing high B_s because the addition of Co decreases the liquidus temperature which facilitates the amorphous phase formation [12,13]. However, noble metal elements such as Co compared to others present in the alloy detract the commercialization. Recent researches suggest that minor C addition is effective to obtain good soft magnetic properties (SMP) in thicker ribbons [14-17]. The enhancement of AFA by C addition is owing to the small size of metalloid atom which disrupts the short-range order, the formation of competition complex phases which impedes the devitrification as well as the approach of deep eutectic point. Nevertheless, the above researches are only based on a small amount of C addition with < 1 at.%. Meanwhile, it is worth mentioning that P plays a crucial role on AFA and SMP of NANOMET alloys as there are previous reports

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revealing that the Cu₃P like clusters are easily formed due to the strong attractive interaction between P and Cu atoms, which possibly act as nucleation sites for α -Fe phase in the as-quenched state and consequently lead to the formation of the hetero-amorphous structure [18,19]. Therefore, with the aim of synthesizing a novel NANOMET-type alloy with higher AFA, good SMP as well as low cost, we developed the Fe_{83.3}Si₄B₈P_{4.x}C_xCu_{0.7} (x = 0, 1, 2, 3, 4) alloys. The influence of C substitution for P on amorphous forming ability, crystallization behavior, microstructure and soft magnetic properties in FeSiBPCCu alloys was investigated.

2. Experimental procedures

Alloy ingots with nominal compositions of Fe_{83.3}Si₄B₈P_{4-x}C_xCu_{0.7} $(x = 0, 1, 2, 3, 4, denoted as C_0 to C_4, respectively)$ were prepared by induction melting the mixtures of Fe (99.99 mass%), Si (99.99 mass%), B (99.5 mass%), Cu (99.99 mass%) and pre-alloyed Fe-P and Fe-C alloy under an argon atmosphere. The alloy ingots were re-melted in a quartz tube with nozzle in the bottom under an argon atmosphere and then sprayed into a rapidly rotating copper wheel under a pressure of 0.02 MPa to obtain amorphous ribbons. The nozzle size is 0.5 mm in diameter and the distance between nozzle and copper wheel is 0.3 mm. Ribbon thickness was controlled by changing the linear wheel speed in a range of 20-50 m/s. The width and thickness were 1 mm and 16-50 µm, respectively. Thermal stability associated with crystallization temperature (T_x) and melting temperature (T_m) of melt-spun alloys was examined by differential scanning calorimetry (DSC) at a heating rate of 0.67 °C/s. The liquidus temperature (T_1) was measured with DSC by cooling the molten alloy samples at a cooling rate of 0.067 °C/s. The melting enthalpy ($\Delta H_{\rm m}$) was determined by the area of the melting peak and, accordingly the melting entropy (ΔS_m) was calculated using the definition of $\Delta S_m = \Delta H_m/T_m$. Melt-spun ribbons were isothermally annealed to develop nanocrystalline alloys. The annealing was carried out by keeping the ribbons in a quartz tube under vacuum atmosphere $(10^{-3} Pa)$, and then pushing the quartz tube into a tubular furnace preheated to annealing temperature (410-530 °C) for 3 min followed by water quenching cooling. Microstructures were identified by X-ray diffraction (XRD) with Cu $K\alpha_1$ radiation and transmission electron microscopy (TEM). TEM samples were prepared by ion milling method (Gatan Inc., PIPS-M691) under liquid nitrogen cooling condition. The volume fraction (V_{crv}) of α -Fe nanocrystals is calculated by equation $V_{\rm crv} = I_{\rm crv}/(I_{\rm crv} + I_{\rm amo})$ where $I_{\rm crv}$ and $I_{\rm amo}$ stand for the integral intensities of diffraction peaks for crystallization phase and amorphous phase, respectively [20]. The values of $V_{\rm cry}$ can be calculated using the Jade 5.0 software by peak fitting. B_s and H_c under a maximum applied field of 800 kA/m and 1 kA/m were measured by a vibrating sample magnetometer (VSM) and a DC B-H loop tracer, respectively. Permeability was measured with an impedance analyzer under a field of 1 A/m. Resistance to DC bias was identified by an impedance analyzer under applied DC field from 1 to 120 A/m. Measurement errors (such as B_s , H_c , etc.) were determined by averaging the measured values of 5 specimens and calculation errors (such as ΔS_m , $V_{\rm cry}$, etc.) were determined by multiple calculations.

3. Results and discussion

Fig. 1 (a) shows the XRD patterns of free surface of $Fe_{83.3}Si_4B_8P_4$. $_xC_xCu_{0.7}$ melt-spun ribbons with 20–21 µm in thickness. It can be seen that there is a sharp crystallization peak at $2\theta = 65^\circ$ existing for $Fe_{83.3}Si_4B_8P_4Cu_{0.7}$ alloy which is related to the (200)-reflection of the crystalline α -Fe phase. However, with increasing C addition, the crystallization peak disappeared and all samples exhibit halo patterns, indicating the formation of amorphous structure. Since surface crystallization behavior is common in high Fe content alloys [21–23], further analysis is taken for the C_0 alloy. Fig. 1 (b) presents the XRD patterns of $Fe_{83.3}Si_4B_8P_4Cu_{0.7}$ alloy in different conditions. The wheel side of

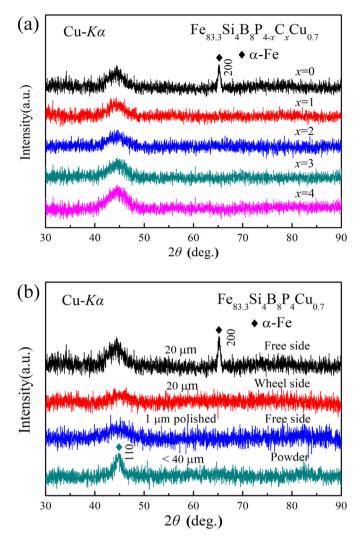


Fig. 1. XRD patterns of (a) free surface of $Fe_{83.3}Si_4B_8P_{4-x}C_xCu_{0.7}$ melt-spun ribbons and (b) $Fe_{83.3}Si_4B_8P_4Cu_{0.7}$ melt-spun ribbons in different conditions.

ribbon sample shows broad peak without appreciable crystallization peaks, which confirms the crystallization only appears on the free side. The in-plane crystalline texture is induced only on the free surface area during quenching, as can be proven by mechanical polishing. As shown in Fig. 1 (b), the ribbon sample polished by 1 µm shows amorphous feature indicating the surface crystallization layer is < 1 µm. It has been revealed that the (200) texture is oriented parallel to the ribbon surface, but there is no preferred orientation along the long ribbon axis [22,23]. In order to confirm this phenomenon, the Fe_{83.3}Si₄B₈P₄Cu_{0.7} powder ground manually in alcohol from the melt-spun ribbon was characterized. It is found that the crystallization peak at $2\theta = 65^{\circ}$ disappears and only a small diffraction peak appears at about $2\theta = 44.5^{\circ}$ for the powders with particle sizes < 40 µm. The result reveals that the (200) texture disappeared when the ribbon is ground into powder, which is well consistent with previous result [22].

Single roller melt-spinning method was used to evaluate the AFA. Ribbon thickness was controlled by changing the wheel speed. The critical thickness and wheel speed were selected as the characteristics of AFA. Fig. 2 shows the XRD patterns of free surface of $Fe_{83.3}Si_4B_8P_{4-x}C_xCu_{0.7}$ (x = 0, 2, 4) melt-spun ribbons with different thickness. For C_0 alloy, only 16–18 µm-thick amorphous ribbon can be prepared with a high wheel speed of 50 m/s. The C-containing alloys exhibit higher AFA and can be made into single amorphous phase at a lower speed (25–35 m/s). Increasing the ribbon thickness to above 20 µm for C_0 and 39 µm for C_2 ribbons results in the appearance of a sharp crystallization

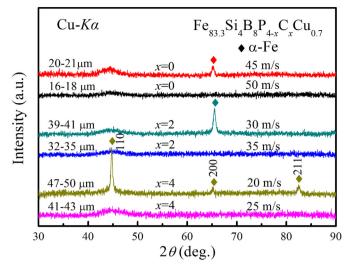


Fig. 2. XRD patterns of free surface of $Fe_{83.3}Si_4B_8P_{4.x}C_xCu_{0.7}$ (x = 0, 2, 4) melt-spun ribbons with different thickness.

peak at $2\theta = 65^{\circ}$, which is associated with the surface crystallization. However, the 47–50 µm-thick C₄ sample shows the Bragg peaks associated with three characteristic diffraction peaks of α -Fe, indicating the precipitation of nanocrystalline α -Fe phase.

Fig. 3 summarizes the amorphous forming ability characterized by critical thickness. The AFA of $Fe_{83.3}Si_4B_8P_4Cu_{0.7}$ alloy is improved significantly by C substitution and the critical thickness increases from 18 µm for C_0 alloy to 43 µm for C_4 alloy. It is noted that the AFA of C_4 alloy is higher than that of NANOMET-type alloys with 34 µm in critical thickness [15]. The result seems to be inconsistent with previous reports which indicate that C addition can effectively increase the AFA, but the optimal C content is ≤ 0.3 at.% and the critical thickness is limited to 25 µm or less [16,17]. The present result shows that even > 2 at.% C additions are beneficial to the formation of amorphous phase. The primary phase is determined as α -Fe (200) phase for the C free and minor C-containing alloys, however it changes into α -Fe phases with different orientation by complete C substitution. The formation of crystalline α -Fe phase with different textures may be beneficial for the AFA due to the competing effect.

Thermodynamic properties of melt-spun ribbons were investigated for further understanding the AFA. The ribbons used for the following measurement are $20-21 \,\mu\text{m}$ in thickness. Fig. 4 (a) shows the DSC

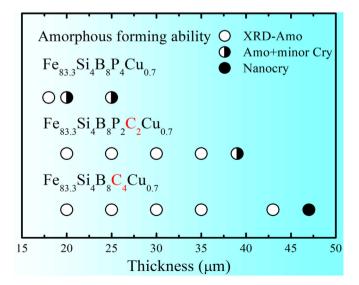


Fig. 3. Amorphous forming ability of $Fe_{83.3}Si_4B_8P_{4-x}C_xCu_{0.7}$ alloys (x = 0, 2, 4).

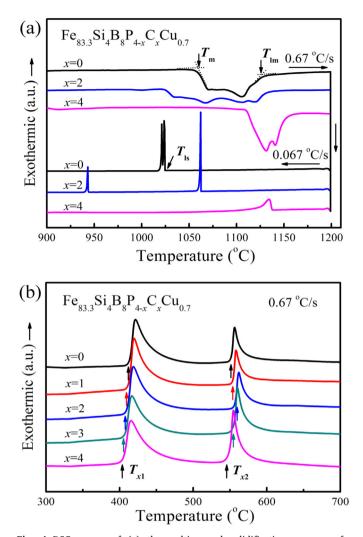


Fig. 4. DSC curves of (a) the melting and solidification processes for $Fe_{83.3}Si_4B_8P_{4-x}C_xCu_{0.7}$ (x = 0, 2, 4) alloys and (b) the crystallization process for $Fe_{83.3}Si_4B_8P_{4-x}C_xCu_{0.7}$ alloys.

curves of the melting and solidification processes for Fe_{83.3}Si₄B₈P₄₋ $_{x}C_{x}Cu_{0.7}$ (x = 0, 2, 4) alloys. The onset and offset temperatures of the melting endothermic event are designated by $T_{\rm m}$ and $T_{\rm lm}$. The onset temperature of the solidification exothermic event is designated by $T_{\rm ls}$. The thermal parameters are listed in Table 1. With 2 at.% C addition, the melting temperature interval between $T_{\rm m}$ and $T_{\rm 1m}$ increases from 70 to 105 °C, combined with the increase in $T_{\rm ls}$, which means poor AFA. However, the $\Delta H_{\rm m}$ decreases clearly, indicating the lower binding energy of the crystalline phases. Meanwhile, the calculated $\Delta S_{\rm m}$ (3.56 J/ mol-K) for C₂ alloy is lower than that of C₀ alloy, which favors the formation of amorphous, as it has been experimentally evidenced that alloys of higher AFA generally have lower $\Delta S_{\rm m}$ [24–26]. $\Delta S_{\rm m}$ has long been recognized as a crucial thermodynamic quantity in determining kinetic behaviors of materials. In addition to decreasing thermodynamic driving forces for crystallization in a liquid-solid transition, low $\Delta S_{\rm m}$ also reduces crystal growth rate and interfacial tension, steepens the profiles of liquidus lines in equilibrium phase diagrams which have an advantage in obtaining deep eutectics therefore favors the glass formation [25,26]. With further increasing C content to 4 at.%, the melting temperature interval decreases obviously from 105 to 50 °C, implying that the composition of the alloy lies in the vicinity of a eutectic point with further increasing C addition. Although the C₄ alloy exhibits a higher T_{ls} due to the substitution of C element with high melting point, which may lead to the increase in eutectic temperature,

Alloys	Thermal properties	roperties					AFA		Magnetic properties	erties			Microstructı	Microstructure parameters	
	T_{x1} (°C)	T _{x2} (°C)	$T_{\rm m}$ (°C)	T_{lm} (°C)	$T_{X1} (\ \ C) T_{X2} (\ \ C) T_{m} (\ \ C) T_{m} (\ \ C) \Delta S_{m} (J/mol-K) E_{p} (kJ/mol) V_{c} (m/s) T (\mu m)$	E _p (kJ/mol)	V _c (m/s)	T (μm)	B _s (T)	$H_{\rm c}$ (A/m) $\mu_{\rm e}$ (1 kHz)	$\mu_{\rm e}$ (1 kHz)	$\mu_{120}/\mu_{\rm e}~(\%)$	D (nm)	D (nm) $V_{\rm cry}$ (%)	N _d (m ⁻³)
${\rm Fe}_{83.3}{\rm Si}_4{\rm B}_8{\rm P}_4{\rm Cu}_{0.7}$	414	552	1058	1128	4.13 ± 0.05	255 ± 5	50	18 ± 1	1.69 ± 0.01	4.3 ± 0.3	$17,380 \pm 500$	44.9 ± 2.8	14 ± 0.5	30.5 ± 2.0	$1.1 imes10^{23}$
$Fe_{83.3}Si_4B_8P_3C_1Cu_{0.7}$	413	554	1021	1132	3.39 ± 0.05	I	I	I	1.82 ± 0.01	5.3 ± 0.4	$11,850 \pm 420$	55.5 ± 3.7	19 ± 0.5	50.4 ± 2.1	I
Fe _{83.3} Si ₄ B ₈ P ₂ C ₂ Cu _{0.7}	410	558	1025	1130	3.56 ± 0.04	269 ± 5	35	35 ± 1	1.84 ± 0.01	4.8 ± 0.3	$13,540 \pm 450$	53.5 ± 3.7	22 ± 0.5	55.6 ± 2.2	$5.2 imes10^{22}$
$Fe_{83.3}Si_4B_8P_1C_3Cu_{0.7}$	408	556	1025	1134	3.09 ± 0.06	I	I	I	1.85 ± 0.01	22.0 ± 2.1	9670 ± 340	72.6 ± 4.9	28 ± 0.6	58.7 ± 2.2	I
${\rm Fe}_{83.3}{\rm Si}_4{\rm B}_8{\rm C}_4{\rm Cu}_{0.7}$	406	550	1110	1160	3.46 ± 0.05	230 ± 5	25	43 ± 1	1.85 ± 0.01	23.3 ± 2.2	6780 ± 240	71.1 ± 4.9	34 ± 0.8	60.6 ± 2.2	$1.5 imes 10^{22}$

Table 1

the ΔS_m is only 3.46 J/mol-K and the alloy shows only one solidification exothermic peak, further implying the approach of a eutectic point that exhibits a higher AFA. In addition, the C atom with small atomic radius can occupy interstitial spaces among the major constituent atoms, leading to the increase in packing density of the liquids. The strong atomic bonding between Fe and C due to the large negative heat of mixing (Fe-C: -50 kJ/mol) [27] can enhance the short-range order of amorphous-forming liquids, which also impedes the devitrification of primary α -Fe phase [28]. Consequently, taking a combination of the lower melting entropy, closer to the eutectic point and micro-alloying effect into consideration, it is concluded that the C₄ alloy possesses a higher AFA with 43 µm in critical thickness.

In order to investigate the magnetic properties, Fe_{83 3}Si₄B₈P₄, $_{x}C_{x}Cu_{0.7}$ alloy ribbons were annealed to develop nanocrystalline alloys. The crystallization behavior was first measured to determine the annealing temperature (T_a) . Fig. 4 (b) shows the DSC curves of Fe_{83.3}Si₄B₈P_{4-x}C_xCu_{0.7} melt-spun ribbons showing the crystallization process. There are two exothermic peaks for the alloys indicating the crystallization process includes two stages. According to former research results, the first onset crystallization temperature (T_{x1}) is associated with the primary crystallization of α -Fe phase while the second one (T_{x2}) corresponds to formation of FeBP compounds [29]. The T_{x1} shows a slight decrease with C increasing, which indicates that the C substitution for P favors the precipitation of initial *a*-Fe phase. Although the temperature difference between T_{x1} and T_{x2} shows no obvious change with the increase in C addition, the large temperature intervals over 138–148 °C are beneficial for the formation of α -Fe phase without nonferromagnetic phase precipitation, which is advantageous for achieving good SMP.

Fig. 5 shows the XRD patterns of $Fe_{83.3}Si_4B_8P_{4.x}C_xCu_{0.7}$ alloys annealed at 450 °C for 3 min. As a result, single α -Fe phase is precipitated obviously from amorphous matrix for the annealed alloys. The intensity of diffraction peak corresponding to the (110) α -Fe phase increases gradually with the increasing C addition, which means higher crystallization. The volume fraction (V_{cry}) of α -Fe nanocrystals is from 30.5% to 60.6% with the increase in *x* from 0 to 4, while the average grain size (*D*) estimated by Scherrer's equation according to the (110) diffraction peak is 15, 19, 21, 28 and 32 nm, respectively, indicating that C substitution for P simultaneously enlarges the V_{cry} and *D*.

The dependence of H_c on T_a for Fe_{83.3}Si₄B₈P_{4-x}C_xCu_{0.7} alloys is shown in Fig. 6. Here AQ stands for H_c of as-quenched alloys. According to the result, H_c first decreases at $T_a = 410$ °C. This decrease is mainly caused by initial stress relief. Then for the alloys with composition of x = 0-2, H_c first increases slightly at $T_a = 430$ °C due to the low

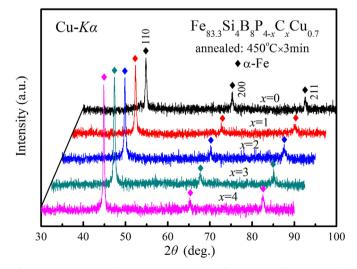


Fig. 5. XRD patterns of $Fe_{83.3}Si_4B_8P_{4\cdot x}C_xCu_{0.7}$ alloys annealed at 450 $^\circ C$ for 3 min.

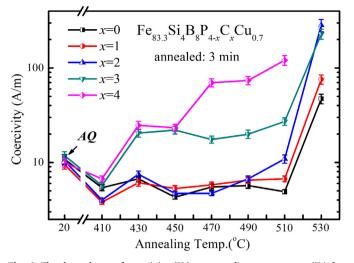


Fig. 6. The dependence of coercivity (H_c) on annealing temperature (T_a) for Fe_{83.3}Si₄B₈P_{4.x}C_xCu_{0.7} alloys.

nucleation rate at low $T_{\rm a}$. With further increased $T_{\rm a}$, $H_{\rm c}$ decreases again and keeps at a low value in a wide $T_{\rm a}$ range between 450 and 510 °C, implying good thermal stability. When $T_{\rm a}$ is increased to 530 °C, $H_{\rm c}$ increases dramatically due to the growth of α -Fe nanocrystals and/or the precipitation of nonferromagnetic phases as 530 °C is very close to the second crystallization temperature. The $H_{\rm c}$ for the C₃ alloy shows a similar variation trend, the difference is that it keeps at a larger value above 17.5 A/m in the same $T_{\rm a}$ range. While for the C₄ alloy, $H_{\rm c}$ has been increased dramatically at 470 °C and cannot be detected by B–H loop tracer at 530 °C as the hysteresis loop exhibits paramagnetic characteristic. According to the results, the optimum $T_{\rm a}$ is fixed as 450 °C.

The hysteresis loops of Fe_{83.3}Si₄B₈P_{4-x}C_xCu_{0.7} alloys annealed at 450 °C for 3 min were then measured as shown in Fig. 7. Accordingly, the B_s of annealed alloys increases obviously from 1.69 to 1.82 T with 1 at.% C substitution and continue to increase with further increasing C content as shown in the upper-left illustration. The increase in B_s is caused by the increasing V_{cry} of α -Fe nanocrystals as the value of B_s is dependent on the magnetization in crystalline and amorphous phases. The B_s of the former is much larger than the latter, therefore the higher the V_{cry} , the superior the B_s [30]. From the inset hysteresis curves detected from *DC* B–H loop tracer, it is known that the alloys with proper

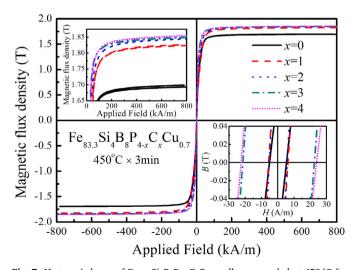


Fig. 7. Hysteresis loops of $Fe_{83.3}Si_4B_8P_{4.x}C_xCu_{0.7}$ alloys annealed at 450 °C for 3 min. The inset is a partial enlargement of that of approaching saturation and hysteresis curves measured by *DC* B–H loop tracer at applied field of 1 kA/m.

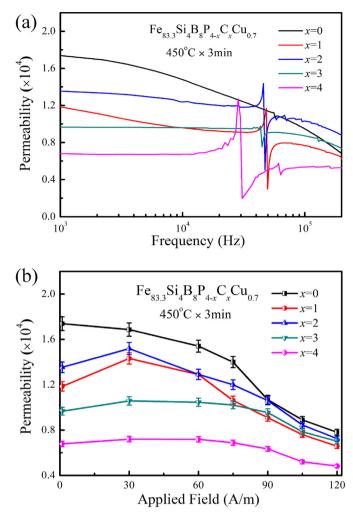


Fig. 8. The dependence of permeability on (a) frequency and (b) applied *DC* field for $Fe_{83.3}Si_4B_8P_{4-x}C_xCu_{0.7}$ alloys annealed at 450 °C for 3 min.

C addition exhibit low H_c of < 5 A/m. It is noted that the annealed C-containing alloys exhibit larger slopes which indicates the higher permeability at an applied field of 1 kA/m.

Considering the relative permeability (μ) under high frequency and applied DC field is an important parameter in application for nanocrystalline alloys [31,32], the dependence of permeability on frequency for Fe83.3Si4B8P4.xCxCu0.7 alloys annealed at 450 °C for 3 min was measured. As shown in Fig. 8 (a), the effective permeability (μ_e) shows a decreasing tendency with increasing C substitution. Although the annealed C₀ alloy possesses the largest μ_e of 17,380, the μ decreases dramatically towards frequency, from which indicating the frequency characteristic is inferior but can be improved by C addition. For the alloy with composition of $x \ge 2$, the μ almost remains unchanged in the wide frequency range of 1–200 kHz. Fig. 8 (b) shows the dependence of permeability on applied DC field for Fe83.3Si4B8P4.xCxCu0.7 alloys annealed at 450 °C for 3 min. It can be seen that the μ decreases gradually towards the density of applied field for the C₀ alloy. With C substitution, there is a slightly increase for μ and the maximum value is reached at 30 A/m, which is quite consistent with the previous researches [32]. For the C₃ and C₄ alloys, the μ exhibits almost no decrease until the amplitude applied field is above 90 A/m. The value of μ under an applied DC field of 120 A/m is 7020 for C₃ and 4820 for C₄ alloy, which is 72.6% and 71.1% of μ_e , respectively, indicating high resistance to DC bias.

Since soft magnetic properties are significantly associated with microstructure, further analysis of microstructure was studied by TEM.

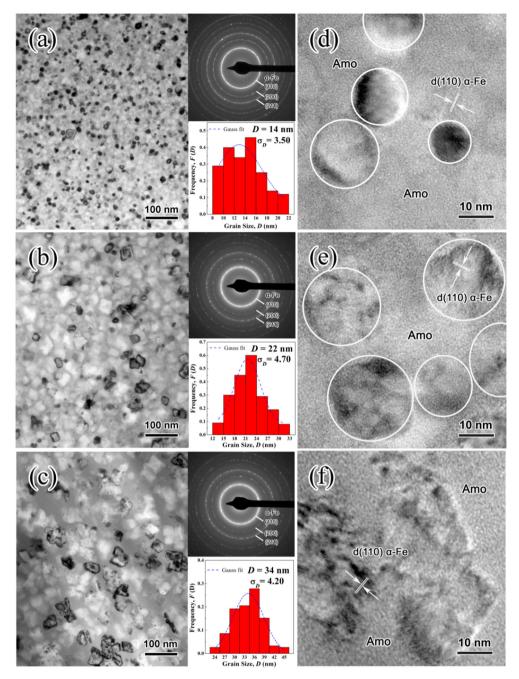


Fig. 9. The bright-field TEM images, selected area electron diffraction (SAED) patterns, grain size distributions and corresponding high resolution-TEM (HRTEM) images of $Fe_{83.3}Si_4B_8P_{4-x}C_xCu_{0.7}$ alloys annealed at 450 °C for 3 min with (a) and (d): x = 0; (b) and (e): x = 2; (c) and (e): x = 4, respectively. The blue dotted lines show Gauss fit. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

Fig. 9 presents the TEM images of Fe_{83.3}Si₄B₈P_{4.x}C_xCu_{0.7} alloys annealed at 450 °C for 3 min with composition of x = 0, 2 and 4, respectively. The bright-field TEM images and SAED patterns (*see* Fig. 9 (a)-(c)) reveal that α -Fe nanocrystalline grains precipitate from amorphous matrix after annealing but the grain size and distribution are quite different. For the annealed C₀ alloy, the precipitated α -Fe grains are relatively small but more uniform. Statistic result reveals that the grain distributes mainly from 8 to 18 nm with 14 nm in average. The number density (*N*_d) of α -Fe grain is around 1.1 × 10²³ m⁻³, which is consistent with previously reported NANOMET alloy [33]. Annealed C₂ alloy also exhibits a uniform nanostructure but the grain size is obviously larger compared with the C₀ alloy. As for the C₄ alloy, the average *D* is as large as 34 nm and *N*_d is decreased from 5.2 × 10²² m⁻³ to 1.5 × 10²² m⁻³. The decrease in *N*_d is resulted from the decreasing P

content, as there have been reports revealing that the coordination of P and Cu atom can provide more heterogeneous nucleation sites to promote the nucleation of α -Fe primary crystals [19,34,35]. Therefore, the alloy with more P-content leads to a higher N_d which greatly lowers the grain size. It has also been reported that P in NANOMET alloy contributes to stabilizing residual amorphous phase hence inhibits the grain growth [36,37]. What is more, the mean square deviation (σ_D) for C₀ alloy is only 3.50, which is smaller than that of C-containing sample ($\sigma_D \geq 4.20$), further reflecting a more uniform nanostructure.

The HRTEM observation (see Fig. 9 (d)-(f)) reveals that the structure and grain shape of α -Fe nanocrystals have certain difference. There exists a large amount of residual amorphous phase for annealed C₀ alloy. However, the increasing C substitution clearly lowers the N_d, resulting in the growth of α -Fe grain thus increasing the V_{cry}. The result

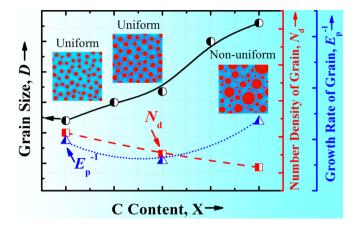


Fig. 10. Microstructure schematics showing the variation trend of grain size (*D*), number density (N_d) and growth rate (E_p^{-1}) of α -Fe grain dependent on C content for FeSiBPCCu nanocrystalline alloys.

is quite consistent with the XRD measurement. Meanwhile, the grain profiles for annealed C_0 and C_2 alloys are circular while the C_4 alloy appears elongated, indicating the existence of orientation growth of α -Fe crystals and large magneto-crystalline anisotropy.

Since crystallization is a solid-state phase transformation controlled by nucleation and growth kinetics, it is important to further understand the reason why proper C substitution for P is beneficial for the formation of nanocrystalline alloys with uniform nanostructure. The crystallization kinetics was studied by using the Kissinger method [38]. The activation energy (E_p) deduced from the first peak temperatures corresponding to the growth of α -Fe grain is calculated as listed in Table 1, the C₂ alloy exhibits the largest E_p of 269 kJ/mol, implying a low growth rate of α -Fe grain. With further increased C addition, the E_p decreases to 230 kJ/mol, indicating the easy growing of α -Fe grain.

Fig. 10 illustrates the microstructure schematics showing the variation trend of *D*, N_d and growth rate (E_p^{-1}) of α -Fe grain dependent on C content for FeSiBPCCu nanocrystalline alloys. As mentioned above, the high N_d of $\sim 1.1 \times 10^{23}$ and relatively low E_p^{-1} make the C free alloy possess ultrafine nanostructure with uniform distribution of α -Fe phase. As for the alloy with minor C addition, although the N_d is less than that of C free alloy, the larger E_p indicates the growth of α -Fe grain is effectively inhibited. The competition between relatively low nucleation density and low growth rate of α -Fe grain also results in a fine nanostructure. However, the excessive C addition not only lowers the N_d but also degrades the thermal stability, which leads to the rapid and non-uniform growth of α -Fe grains, thereby deteriorating the SMP.

Table 1 summarizes the thermal properties, AFA, SMP and microstructure parameters of Fe_{83.3}Si₄B₈P_{4-x}C_xCu_{0.7} alloys annealed at 450 °C for 3 min. The Fe_{83.3}Si₄B₈P₂C₂Cu_{0.7} nanocrystalline alloy simultaneously possesses high B_s of 1.84 T and low H_c of 4.8 A/m. Though H_c is slightly larger than C₀ alloy, the obviously higher V_{cry} with ultrafine nanostructures and homogenous distribution lead to the strengthening of exchange-coupling interaction, which not only effectively superiors the B_s, but also reduces effective magneto-crystalline anisotropy < K > thereby decreasing the H_c as it has been proved by the random anisotropy model which suggests that H_c is directly related to < K > by

$$H_{c} = p_{c} \frac{\langle K \rangle}{J_{s}} \approx p_{c} \frac{K_{1}^{4} D^{6}}{J_{s} A^{3}}$$
(1)

where p_c is dimensionless pre-factor of the order of unity, K_1 denotes the local magneto-crystalline anisotropy, J_s is the average saturation polarization and A is the exchange stiffness. The suppression of magneto-crystalline anisotropy requires that the randomly oriented grains are ferromagnetically coupled by exchange interaction. Consequently, if the exchange interaction is enhanced, the < K > will be effectively averaged out and SMP will be improved [39,40]. It is interesting that the C₁ alloy possesses a lower *D* of 19 nm but it exhibits a larger H_c of 5.3 A/m compared with the C₂ alloy. This phenomenon can also be explained by the grain coupling effect as the V_{cry} is 50.4% for the C₁ alloy, which indicates a weaker exchange coupling between crystallites compared with that of the C₂ alloy. As for the frequency and *DC* superposition performance, it has been confirmed by the previous work that resistance to *DC* bias is related to the arrangement of magnetic domain structure [41]. The microstructure observation has suggested that the alloy with a higher C content exhibits comparatively larger magneto-crystalline anisotropy, which means the domain rearrangement is harder under applied field, therefore improves the resistance to applied *DC* field. The combination of high B_s and high resistance to *DC* bias promises potential material in high magnetic field amplitude and *DC* superposition application.

4. Conclusion

In conclusion, the influence of C substitution for P on amorphous forming ability, crystallization behavior, microstructure and soft magnetic properties in Fe_{83.3}Si₄B₈P_{4-x}C_xCu_{0.7} alloy system was investigated. The results show that C substitution inhibits the precipitation of α -Fe phase in the quenching process. The formation of competing crystalline α -Fe phases with different textures is beneficial for the AFA. B_s increases with increasing C substitution due to the increase in V_{cry} of α -Fe nanocrystals, while excessive C addition leads to larger H_c but improves the resistance to applied *DC* field. The Fe_{83.3}Si₄B₈P₂C₂Cu_{0.7} nanocrystalline alloy is developed with relatively higher AFA with 35 µm in critical thickness, high B_s of 1.84 T, low H_c of 4.8 A/m, high μ_e of 13,540, as well as better frequency characteristic and resistance to *DC* bias. The results can provide guidance for synthesizing novel soft magnetic materials with high AFA and good SMP.

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